

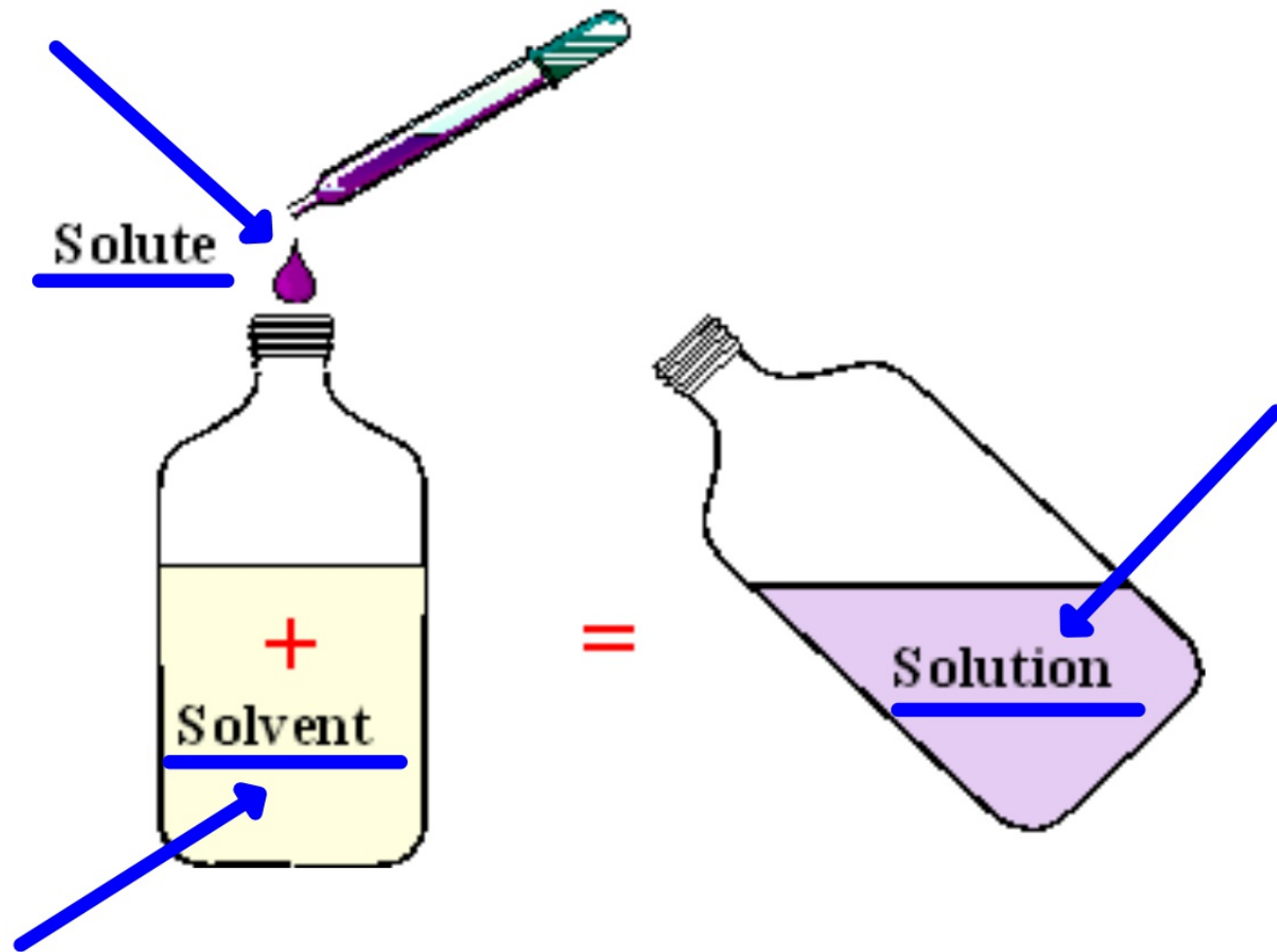
Solubility

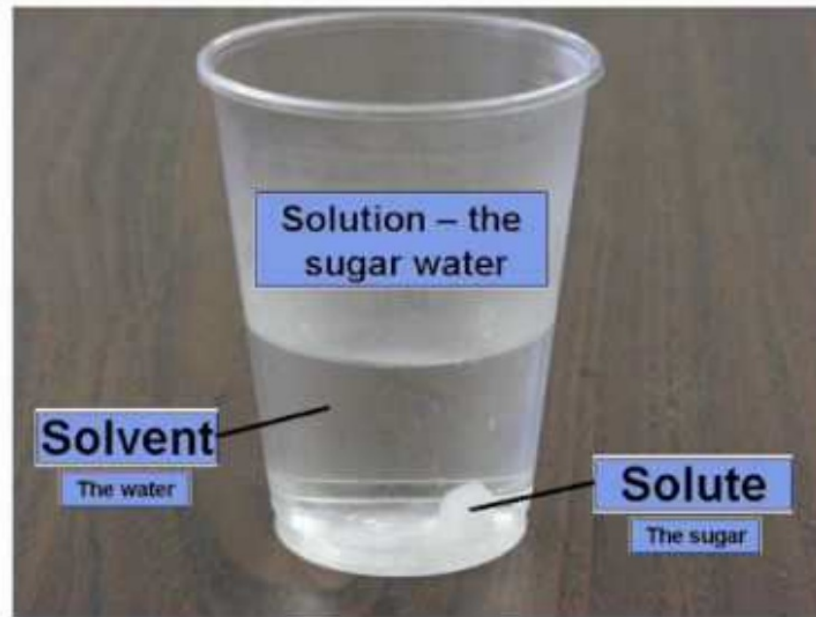
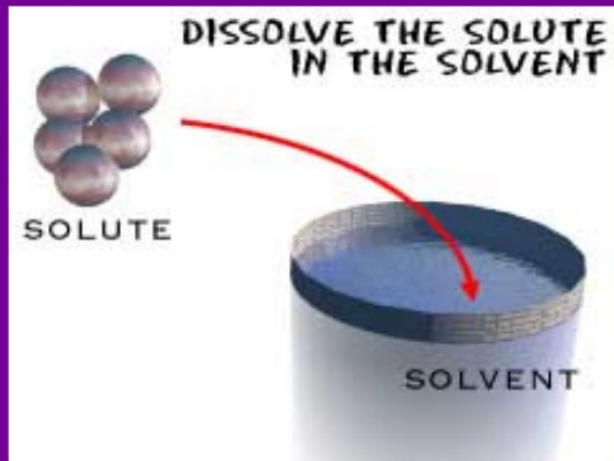


We can determine
if a solid, liquid, or gas
can dissolve in another
substance.

Using our sense:

Sight 







Solubility: A physical property. The amount of solute (ex. Salt) that can be dissolved by the solvent (ex. Water).

Solvent: A substance (ex. water) that dissolves a solute (ex. salt).

Solute: The solute (ex. salt) gets dissolved in a solvent (ex. water), to create a solution.



Solution: A HOMOGENEOUS MIXTURE



SOLVENT

VS.

SOLUTE

How can we remember this?

