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5th grade science

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Density

Using the formulas for mass, volume & density





A brick has a mass of 100 g and a volume of 25 cm³. What is the density of the brick?

The density of gold is 19.3 g/cm³. If I have a nugget with a volume of 10 cm³, what is the mass of the cube?



The amount of matter in an object. The amount of "stuff" is measured in grams.

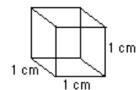
The amount of matter in a given space or volume. g/mL or g/cm³

How to solve a word problem:

- 1. Read the word problem carefully.
- 2. Determine what is being asked for.
- 3. Write the formula and plug in the known values.
- 4. Calculate and solve for the unknown value.
- 5. Write the answer and corresponding unit.

My paperweight has a mass of 50 g and a density of 2.5 g/cm³. How much space does it take up?

Volume of 1 cubic (cm³) centimeter



 $1 \text{ mL} = 1 \text{ cm}^3$



The amount of space an object takes up. We use cm³ or mL to measure volume.