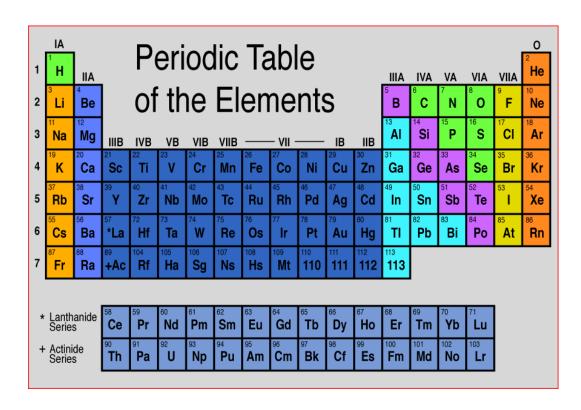
Classifying Matter: Elements, Compounds, and Mixtures

www.middleschoolscience.com 2009

Elements

• Pure substance that cannot be separated into simpler substance by physical or chemical means.



Compounds

Pure substance composed of two or more different elements joined by chemical bonds.

- Made of elements in a specific ratio that is always the same
- Has a chemical formula
- Can only be separated by chemical means, not physically

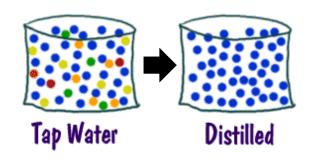






Mixtures

- A combination of two or more pure substances that are not chemically combined.
- substances held together by *physical forces*, not chemical
- No chemical change takes place
- Each item retains its properties in the mixture
- They can be separated physically



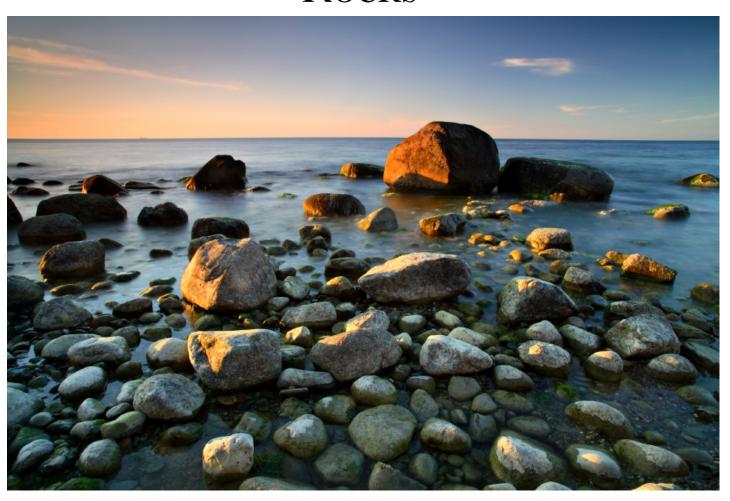


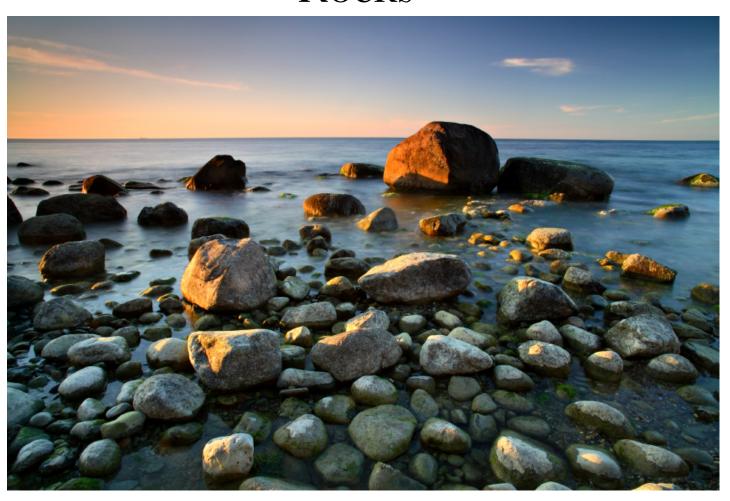
Can you identify the following?

Leaders: Each Group Needs One Whiteboard and One Marker.

Review:

- An **element** contains just one type of atom.
- A **compound** contains two or more different atoms joined together.
- A **mixture** contains two or more different substances that are only physically joined together, not chemically.
 - A mixture can contain both elements and compounds.





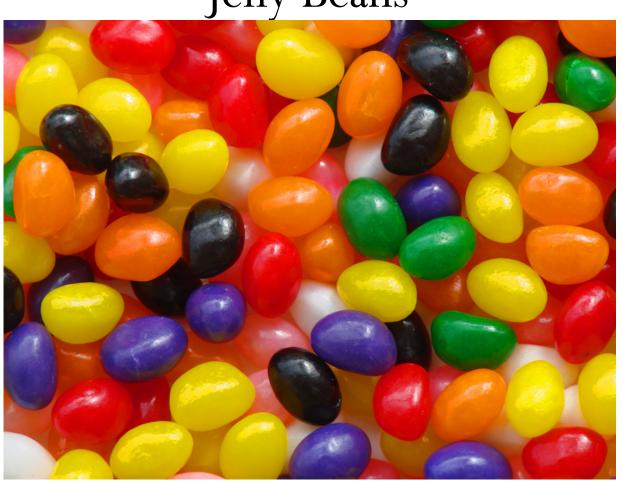


Copper

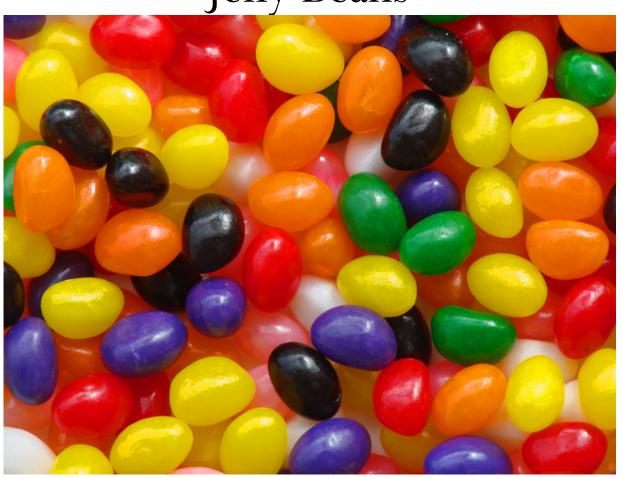




Jelly Beans



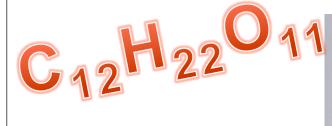
Jelly Beans



Element, Compound, or Mixture? Table Sugar

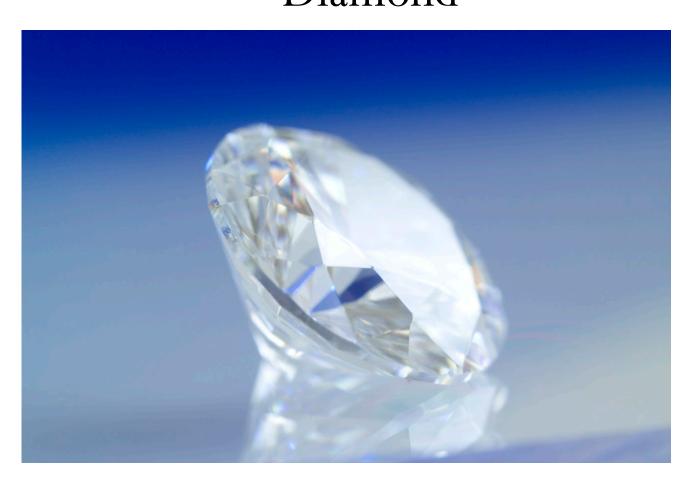


Table Sugar





Element, Compound, or Mixture? Diamond





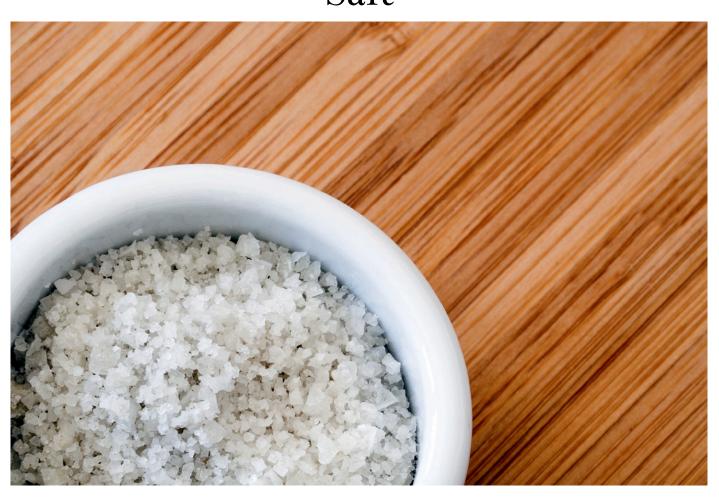
Diamond





Tea





NaCl Salt



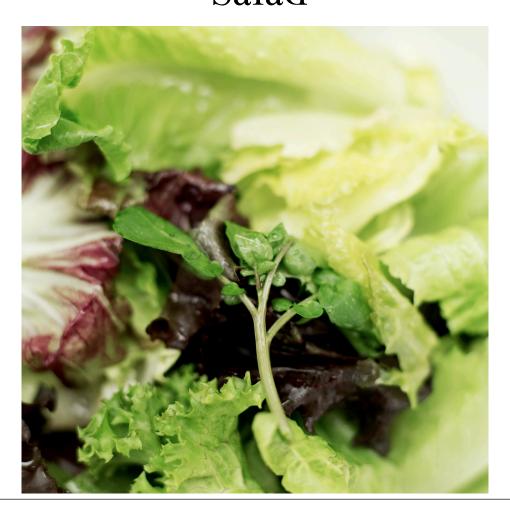
Neon Gas



Neon Gas







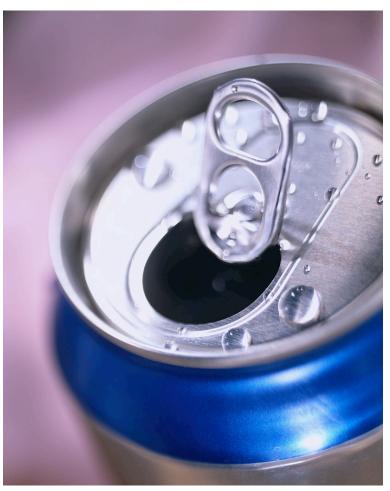
Pure Water



Pure Water

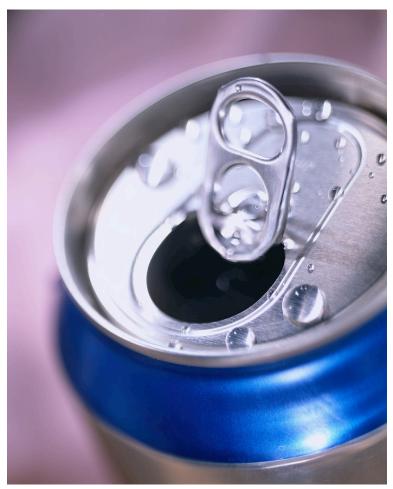


Aluminum



Al

Aluminum



Lemonade



Lemonade



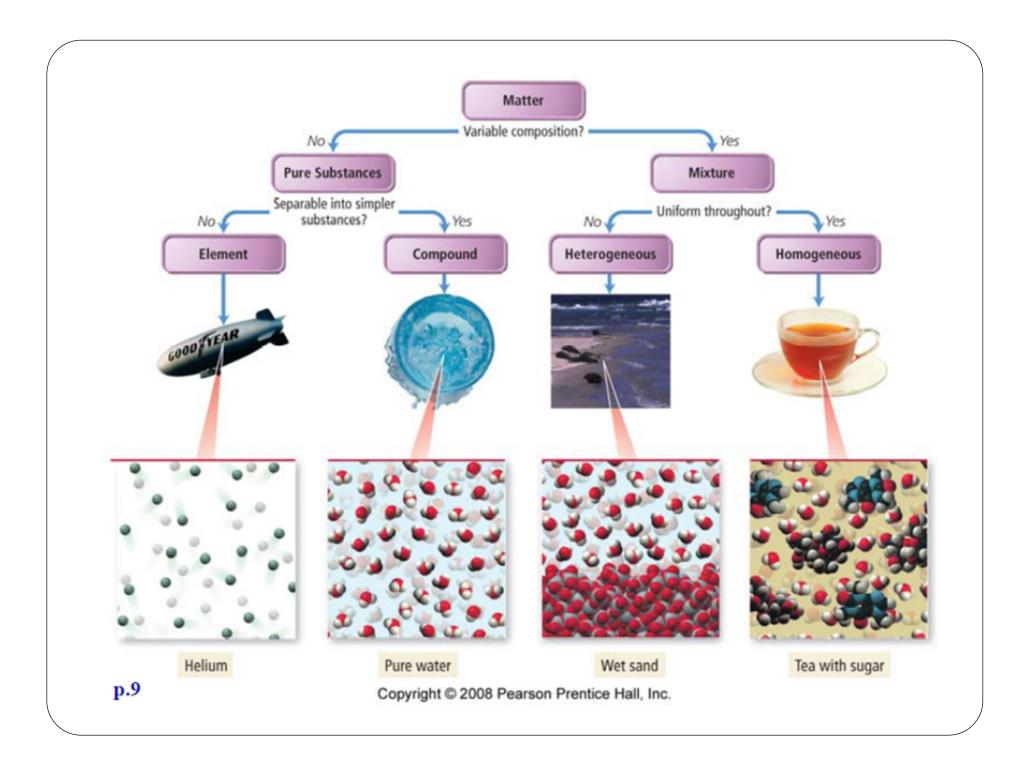






Sand

Sand



Notes

• Detailed notes are located at:

http://www.middleschoolscience.com/elements-compoundsmixtures-notes-isn.pdf

• Flow Chart:

http://www.middleschoolscience.com/matter-flow-chartisn.pdf